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A [2+2+2]-cyclotrimerization approach for the synthesis of enantiopure isochromans using a carbohydrate derived dialkyne template

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Abstract

An easy access to enantiopure isochromans through cross alkyne trimerization of a glucose derived dialkyne was developed. One of the synthesized isochromans was converted into a novel tricyclic nucleoside by simple transformations. © 2007 Elsevier Ltd. All rights reserved.

Keywords: Isochroman; [2+2+2]-Cyclotrimerization; Wilkinson's catalyst; Vorbrüggen reaction; Tricyclic nucleoside

Designing effective routes to construct complex cyclic structures through organo transition-metal catalyzed reactions has been recognized as an attractive strategy for delivering molecular diversity.^{[1](#page-2-0)} Integrated with transitionmetal catalyzed reactions, sugar templates have been well deployed to address the synthesis of a variety of complex natural product skeletons.² Amongst the many other metal catalyzed reactions which have been explored on sugar templates, catalytic [2+2+2]-alkyne cyclotrimerizations are important as they deliver highly functionalized aromatic rings appended with sugar rings. 3 An early example in this context is an expedient total synthesis of the spirocyclic C-arylglycoside framework closely related to the papulacandins by McDonald et al. utilizing a rhodium(I) catalyzed $[2+2+2]$ -cyclotrimerization^{[4](#page-2-0)} which has been well explored with various sugar derivatives. 5 Considering the simplicity of the $[2+2+2]$ -alkyne cyclotrimerization combined with our current interest in exploiting carbohydrates for constructing useful molecular diversity, 6 herein, we report the synthesis of enantiomeric tricyclic molecular skeletons consisting of isochroman units (Fig. 1).

Isochroman $(1, 3, 4$ -dihydro-1H-benzo $[c]$ pyran) is a structural unit found in some important biologically active

Fig. 1. Key [2+2+2]-cyclotrimerization approach for enantiopure isochromans.

compounds, commercially available drugs and cosmetics.^{[7](#page-2-0)} Several synthetic 6,7-dimethoxyisochromans and their 1-arylated analogues have been disclosed as new investigational drugs with a wide range of activities such as analgesic, muscle relaxant, antidepressant, anti-inflamma-tory, antihistaminic, anticoagulant and antihypertensive.^{[8](#page-2-0)}

Kakimoto et al. prepared a series of tricyclic isochroman derivatives 2 from glucose and evaluated their herbicidal

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activities.^{[9](#page-3-0)} A number of synthetic methods involving electrophilic reactions mediated by Lewis acids, radical and carbanion-mediated annulations, and cycloadditions have been reported for the synthesis of isochroman units.^{[7](#page-2-0)} The oxa-Pictet–Spengler condensation is a widely used method for the preparation of isochroman units.^{7b} An expedient approach for sugar annulated isochromans by Martin et al. using intramolecular Friedel–Crafts cyclization has been reported.¹⁰ Recently, Kaliappan and Ravikumar reported an intramolecular enyne-metathesis and subsequent trapping of the resulting dienes by various quinones for the synthesis of tetracyclic derivatives containing an isochroman ring.^{[11](#page-3-0)}

Our intended strategy is described in [Figure 1.](#page-0-0) The synthesis of the key diyne 3 started with the propargylation of glucose diacetonide 4 to procure the propargyl ether 5 .^{[12](#page-3-0)} Selective monoacetonide hydrolysis of 5 followed by sodium periodate mediated cleavage and subsequent Ohira–Bestmann alkynylation 13 of the intermediate aldehyde provided 3 in 49% overall yield (Scheme 1). With the fully elaborated diyne framework 3 in place, cyclotrimerization was attempted with 2-butyne-1,4-diol (6). After a careful examination of some of the catalysts reported for

Scheme 1. Reagents and conditions: (a) NaH, propargyl bromide, THF, 0 °C to rt, 8 h, 89%; (b) (i) 0.8% H₂SO₄, MeOH, rt, 24 h; (ii) NaIO₄-silica gel, CH₂Cl₂, rt, 1 h; (iii) (MeO)₂P(=O)C(=N₂)COCH₃, methanol, K₂CO₃, rt, 7–9 h, 49% for three steps; (c) $Rh(PPh₃)₂Cl$, toluene–ethanol (4:1), 80 °C, 8 h, 61%.

[2+2+2]-alkyne cyclotrimerization, we found that Wilkinson's catalyst afforded the tricyclic derivative 7 in good yields.^{5d} The spectral and analytical data of 7 were in accordance with the assigned structure.^{[14](#page-3-0)} For example, in the ¹H NMR spectrum of 7, the two aromatic-H appeared

as singlets at δ 6.99 and δ 7.24 ppm. The characteristic $C(1)$ –H and $C(2)$ –H of the furanose ring appeared as doublets at δ 5.92 and 4.70 ppm ($J_{12} = 3.8$ Hz), respectively. C(4)–H appeared downfield (δ 4.89 ppm) as a doublet with $J = 2.3$ Hz. The observed cross peaks in the NOESY spectrum between C(4)–H and the aromatic ring proton at δ 7.24 ppm were indicative of a possible anisotropic effect of the furanose ring oxygen, which helped in assigning the ratios of the regiomeric compounds resulting from unsymmetric alkynes. The presence of two methyleneoxy groups at δ 62.7 and 66.9 ppm in the ¹³C NMR further confirmed the assigned structure.

To illustrate the flexibility of our strategy, various alkynes were employed in the [2+2+2]-cyclotrimerization and the results are summarized in [Table 1.](#page-1-0) With acetylene, the cyclotrimerization reaction proceeded effectively at 80 °C in a sealed tube to afford 8 in good yield.^{[15](#page-3-0)} Unsymmetrical alkynes such as phenylacetylene and hexadec-1 yne gave inseparable regiomeric mixtures in moderate to good yields [\(Table 1\)](#page-1-0).

Having established the feasibility of the $[2+2+2]$ -cyclotrimerization for the synthesis of sugar annulated isochromans, we next applied isochroman 8 for the synthesis of modified nucleoside 19. As shown in Scheme 2, isochroman 8 was subjected to acid catalyzed acetonide hydrolysis using acetic acid followed by acetylation in dichloromethane, acetic anhydride and $Et₃N$ to afford a 1:1 anomeric mixture of 17. Treatment of 17 with uracil under modified Vorbrüggen^{[16](#page-3-0)} conditions afforded the protected nucleoside 18.^{[17](#page-3-0)} Subjecting 18 to Zemplen's deacetylation gave the tricyclic nucleoside 19. The structure of 19^{18} 19^{18} 19^{18} was established with the help of COSY and NOESY spectra that confirmed the assigned β -configuration for 19 beyond doubt.

To conclude, a general synthesis of chiral isochromans via [2+2+2]-cyclotrimerization of a sugar template as the key reaction was reported. One of the derived isochromans was converted into a tricyclic nucleoside by simple synthetic manipulations. Considering the importance of con-

Scheme 2. Reagents and conditions: (a) (i) 60% AcOH, reflux, 2 h, 92%; (ii) Ac₂O, Et₃N, DMAP, CH₂Cl₂, rt, 1 h, 94%; (b) uracil, N,Obis(trimethylsilyl)acetamide (BSA), TMSOTf, CH₃CN, 50 °C, 2 h, 79%; (c) NaOMe, MeOH, rt, 20 min, 96%.

formationally restricted nucleosides as antiviral agents and as potential antisense therapeutic and diagnostic agents, the results from the present investigation could be further explored for a strategic construction of these molecular skeletons. Work in this direction is ongoing in our laboratory.

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References and notes

- 1. (a) Schreiber, S. L. Science 2000, 287, 1964–1969; (b) Schreiber, S. L.; Nicolaou, K. C.; Davies, K. Chem. Biol. 2002, 9, 1–2; (c) Tan, D. S. Nat. Chem. Biol. 2005, 1, 74–84; (d) Walsh, D. P.; Chang, Y.-T. Chem. Rev. 2006, 106, 2476–2530.
- 2. (a) Sinou, D.; Bedjeguelal, K. J. Carbohydr. Chem. 2001, 20, 335–357; (b) Sinou, D.; Bedjeguelal, K. Eur. J. Org. Chem. 2000, 65, 4071– 4077; (c) Weyershausen, B.; Dötz, K. H.; Nieger, M. J. Organomet. Chem. 2000, 602, 37–44; (d) Weyershausen, B.; Dötz, K. H.; Nieger, M. J. Org. Chem. 1999, 64, 4206–4210; (e) Weyershausen, B.; Dötz, K. H. Eur. J. Inorg. Chem. 1999, 1057–1066; (f) Kubota, H.; Lim, J.; Depew, K. M.; Schreiber, S. L. Chem. Biol. 2002, 9, 265–276.
- 3. (a) Yamamoto, Y. Curr. Org. Chem. 2005, 9, 503–519; (b) Ferrier, R. J.; Hoberg, J. O. Adv. Carbohydr. Chem. Biochem. 2003, 58, 55–119; (c) Panigot, M. J.; Kim, S.-U.; Arnold, M. W.; Bailey, A.; Bailey, D.; Faulkner, J. L.; Middleton, J. Toward the synthesis of C-glycoside dendrimers; American Chemical Society, Polymer Preprints, Division of Polymer Chemistry, 2000; (d) Roy, R.; Das, S. K.; Dominique, R.; Corazon Trono, M.; Hernández-Mateo, F.; Santoyo-González, F. Pure Appl. Chem. 1999, 71, 565–571.
- 4. McDonald, F. E.; Zhu, H. Y. H.; Holmquist, C. R. J. Am. Chem. Soc. 1995, 117, 6605–6606.
- 5. (a) Yamamoto, Y.; Saigoku, T.; Ohgai, T.; Nishiyama, H.; Itoh, K. Chem. Commun. 2004, 10, 2702–2703; (b) Yamamoto, Y.; Saigoku, T.; Nishiyama, H.; Ohgai, T.; Itoh, K. Org. Biomol. Chem. 2005, 3, 1768–1775; (c) Yamamoto, Y.; Hashimoto, T.; Hattori, K.; Kikuchi, M.; Nishiyama, H. Org. Lett. 2006, 8, 3565–3568; (d) Novak, P.; Pohl, R.; Kotora, M.; Hocek, M. Org. Lett. 2006, 8, 2051–2054.
- 6. (a) Gurjar, M. K.; Nayak, S.; Ramana, C. V. Tetrahedron Lett. 2005, 46, 1881–1884; (b) Ramana, C. V.; Mallik, R.; Gonnade, R. G.; Gurjar, M. K. Tetrahedron Lett. 2006, 47, 3649–3652; (c) Ramana, C. V.; Patel, P.; Gonnade, R. G. Tetrahedron Lett. 2007, 48, 4771–4774.
- 7. (a) Markaryan, E. A.; Samodurova, A. G. Russ. Chem. Rev. 1989, 58, 479–493; (b) Larghi, E. L.; Kaufman, T. S. Synthesis 2006, 187–220; (c) Arimitsu, S.; Hammond, G. B. J. Org. Chem. 2006, 71, 8665– 8668.
- 8. (a) McCall, J. M.; McCall, R. B.; TenBrink, R. E.; Kamdar, B. V.; Humphrey, S. J.; Sethy, V. H.; Harris, D. W.; Daenzer, C. J. Med. Chem. 1982, 25, 75–81; (b) TenBrink, R. E.; Bergh, C. L.; Duncan, J. N.; Harris, D. W.; Huff, R. M.; Lahti, R. A.; Lawson, C. F.; Lutzke, B. S.; Martin, I. J.; Rees, S. A.; Schlachter, S. K.; Sih, J. C.; Smith, M. W. J. Med. Chem. 1996, 39, 2435–2437; (c) Unterhalt, B.; Jöstingmeier, R.; Sanatgar, A. Pharmazie 1997, 52, 186–189; (d) Bury, P. S.; Christiansen, L. B.; Jacobsen, P.; Jorgensen, A. S.; Kanstrup, A.; Narum, L.; Bain, S.; Fledelius, C.; Gissel, B.; Hansen, B. S.; Korsgaard, N.; Thorpe, S. M.; Wassermann, K. Bioorg. Med. Chem. 2002, 10, 125–145; (e) Liu, J.; Birzin, E. T.; Chan, W.; Yang, Y. T.; Pai, L.-Y.; DaSilva, C.; Hayes, E. C.; Mosley, R. T.; DiNinno, F.; Rohrer, S. P.; Schaeffer, J. M.; Hammond, M. L. Bioorg. Med. Chem. Lett. 2005, 15, 715–718; (f) Suzuki, T.; Tanemura, K.; Horaguchi, T.; Kaneko, K. Tetrahedron 2006, 62, 3739–3751; (g) Mohr, P.; Decker,

M.; Enzensperger, C.; Lehmann, J. J. Med. Chem. 2006, 49, 2110– 2116.

- 9. Kakimoto, T.; Koizumi, F.; Hirase, K.; Banba, S.; Tanaka, E.; Arai, K. Pest Manag. Sci. 2004, 60, 493–500.
- 10. (a) Martin, O. R. Carbohydr. Res. 1987, 171, 211–222; (b) Martin, O. R.; Hendricks, C. A. V.; Deshpande, P. P.; Cutler, A. B.; Kane, S. A.; Rao, S. P. Carbohydr. Res. 1990, 196, 41–58; (c) Kulkarni, S. S.; Liu, Y.-H.; Hung, S.-C. J. Org. Chem. 2005, 70, 2808–2811.
- 11. Kaliappan, K. P.; Ravikumar, V. Org. Biomol. Chem. 2005, 3, 848–851.
- 12. (a) Rochet, P.; Vatele, J.-M.; Gore, J. Synthesis 1994, 795–799; (b) Roy, A.; Sahabuddin, S. K.; Achari, B.; Mandal, S. B. Tetrahedron 2005, 61, 365–371.
- 13. (a) Ohira, S. Synth. Commun. 1989, 19, 561–564; (b) Roth, G. J.; Liepold, B.; Müller, S. G.; Bestmann, H. J. Synlett 1996, 521-522.
- 14. Spectral data of compound 7: $[\alpha]_D^{25}$ 28.0 (c 1.2, CHCl₃). IR (CHCl₃) v: 3401, 3016, 2933, 1734, 1513, 1452, 1438, 1376, 1247, 1216, 1163, 1118, 1102, 1080, 1017, 898, 857, 830, 755, 667 cm⁻¹. ¹H NMR $(200 \text{ MHz}, \text{CDCl}_3)$ δ : 1.35 (s, 3H), 1.58 (s, 3H), 3.90 (br s, 2H), 4.09 $(d, J = 2.3 \text{ Hz}, 1\text{ H}), 4.47-4.60 \text{ (m, 5H)}, 4.67 \text{ (d, } J = 3.8 \text{ Hz}, 1\text{ H}), 4.70 \text{ }$ $(d, J = 15.1 \text{ Hz}, 1\text{H})$, 4.89 $(d, J = 2.3 \text{ Hz}, 1\text{H})$, 5.92 $(d, J = 3.8 \text{ Hz}, 1\text{H})$), 6.99 (s, 1H), 7.24 (s, 1H). ¹³C NMR (50 MHz, CDCl₃) δ : 26.1 (q), 26.7 (q), 62.7 (t), 66.9 (t), 73.1 (d), 79.8 (d), 84.4 (d), 104.9 (d), 111.7 (s), 124.6 (d), 128.4 (s), 130.9 (d), 132.1 (d), 134.4 (s), 138.2 (s), 140.2 (s) ppm. ESI-MS m/z : 301.26 (100%), 309.36 (2.3%, $[M+1]$ ⁺), 331.32 $(89.2\%, \text{ [M+Na]}^+), 347.30 (3.5\%, \text{ [M+K]}^+).$ Anal. Calcd for $C_{16}H_{20}O_6$: C, 62.33; H, 6.54. Found: C, 62.12; H, 6.68.
- 15. Spectral data of compound 8: $[\alpha]_D^{25}$ 21.6 (c 1.5, CHCl₃). IR (CHCl₃) v: 3018, 2927, 1458, 1376, 1216, 1164, 1116, 1091, 1020, 920, 754, 666 cm⁻¹. ¹H NMR (200 MHz, CDCl₃): δ 1.28 (s, 3H), 1.50 (s, 3H), 4.06 (d, $J = 2.4$ Hz, 1H), 4.55–4.75 (m, 3H), 4.87 (d, $J = 2.3$ Hz, 1H), 5.89 (d, $J = 3.91$ Hz, 1H), 6.94–7.01 (m, 1H), 7.17–7.25 (m, 2H), 7.35– 7.41 (m, 1H). ¹³C NMR (50 MHz, CDCl₃) δ : 26.3 (g), 26.9 (g), 67.4 (t), 73.5 (d), 80.1 (d), 84.7 (d), 105.2 (d), 111.6 (s), 124.2 (d), 127.4 (d),

128.7 (d), 129.5 (s), 130.7 (d), 134.8 (s) ppm. ESI-MS: m/z 266.32 (35.9%, $[M+NH_4]^+$), 271.29 (100%, $[M+Na]^+$). Anal. Calcd for C14H16O4: C, 67.73; H, 6.50. Found: C, 67.44; H, 6.89.

- 16. (a) Niedballa, U.; Vorbrüggen, H. J. Org. Chem. 1974, 39, 3654–3660; (b) Vorbrüggen, H.; Krolikewiez, K.; Bennua, B. Chem. Ber. 1981, 114, 1234–1255; (c) Vorbrüggen, H.; Höfle, G. Chem. Ber. 1981, 114, 1256–1268.
- 17. Spectral data of compound 18: $[\alpha]_D^{25} 8.6$ (c 1.0, CHCl₃). IR (CHCl₃) v: 3032, 2927, 1750, 1682, 1461, 1371, 1320, 1268, 1225, 1108, 1066, 909, 78, 761, 734, 667, 602 cm⁻¹. ¹H NMR (200 MHz, CDCl₃) δ : 2.17 (s, 3H), 4.18 (d, $J = 2.6$ Hz, 1H), 4.71 (d, $J = 15.16$ Hz, 1H), 4.86 (d, $J = 15.16$ Hz, 1H), 4.95 (d, $J = 2.5$ Hz, 1H), 5.22 (d, $J = 1.6$ Hz, 1H), 5.60 (dd, $J = 2.1$, 8.2 Hz, 1H), 6.15 (d, $J = 1.6$ Hz, 1H), 7.11 (dd, $J = 2.1$, 8.2 Hz, 1H), 7.32–7.48 (m, 4H), 9.21 (br s, 1H). ¹³C NMR $(50 MHz, CDCl₃)$ δ : 20.7 (q), 67.3 (t), 75.4 (d), 78.8 (d), 81.7 (d), 89.3 (d), 102.5 (d), 124.4 (d), 127.82 (d), 127.9 (s), 129.4 (d), 130.6 (d), 134.1 (s), 140.5 (d), 150.2 (s), 163.1 (s), 169.2 (s) ppm. ESI-MS: m/z 344.09 (5.11%, [M]+), 367.02 (100%, [M+Na]+). Anal. Calcd for C17H16N2O6: C, 59.30; H, 4.68; N, 8.14. Found: C, 59.03; H, 4.90; N, 7.97.
- 18. Spectral data of compound 19: $[\alpha]_D^{25}$ 60.4 (c 1.9, CHCl₃). IR (CHCl₃) v: 3383, 3218, 3108, 3068, 2923, 2844, 2252, 1775, 1695, 1493, 1464, 1408, 1393, 1373, 1360, 1322, 1265, 1206, 1114, 1097, 1076, 1059, 999, 960, 908, 877, 813, 788, 733, 649 cm⁻¹. ¹H NMR (200 MHz, CDCl₃) δ : 4.22 (d, J = 1.9 Hz, 1H), 4.52 (s, 1H), 4.64 (d, J = 15.1 Hz, 1H), 4.72 (d, $J = 15.1$ Hz, 1H), 5.24 (d, $J = 1.9$ Hz, 1H), 5.48 (d, $J = 8.0$ Hz, 1H), 5.65 (s, 1H), 5.89 (s, 1H), 7.1 (dd, $J = 2.3$, 8.1 Hz, 1H), 7.15 (d, $J = 8.03$ Hz, 1H), 7.34–7.36 (m, 2H), 7.46 (dd, $J = 2.3$, 8.1 Hz, 1H), 10.61 (br s, 1H). ¹³C NMR (50 MHz, CDCl₃) δ : 66.9 (t), 77.1 (d), 80.1 (d), 80.7 (d), 94.2 (d), 101.1 (d), 124.3 (d), 127.8 (d), 128.8 (s), 129.2 (d), 130.8 (d), 134.3 (s), 140.7 (d), 151.1 (s), 164.2 (s) ppm. ESI-MS: m/z 325.01 (100%, [M+Na]⁺), 340.99 (7.3%, [M+K]⁺). Anal. Calcd for C₁₅H₁₄N₂O₅: C, 59.60; H, 4.67; N, 9.27. Found: C, 59.33; H, 4.92; N, 8.99.